

STRUCTURES OF SOME MOENOMYCIN ANTIBIOTICS - INHIBITORS OF PEPTIDOGLYCAN BIOSYNTHESIS

Jürgen Scherkenbeck^a, Aranka Hiltmann^a, Kurt Hobert^a, Wassja Bankova^a, Torsten Siegels^a,
Manfred Kaiser^a, Dietrich Müller^a, Hans Jürgen Veith^b, Hans-Wolfram Fehlhaber^c,
Gerhard Seibert^c, Astrid Markuse^c, Michael Limbert^c,
Gerhard Huber^c, Dirk Böttger^c, Andreas Stark^c, Shuji Takahashi^d,
Yveline van Heijenoort^e, Jean van Heijenoort^e, and Peter Welzel^{a*}

^aFakultät für Chemie der Ruhr-Universität, Postfach 102148, D-4630 Bochum (Germany)

^bInstitut für Organische Chemie der Technischen Hochschule Darmstadt
Petersenstr 22, D-6100 Darmstadt (Germany)

^cHoechst AG, Postfach 800320, D-6230 Frankfurt 80 (Germany)

^dSankyo Co, Ltd, Fermentation Research Laboratories,
2-58, Hiromachi 1-chome, Shinagawa-ku, Tokyo (Japan)

^eBiochimie Moléculaire et cellulaire, Université Paris-Sud, Orsay (France)

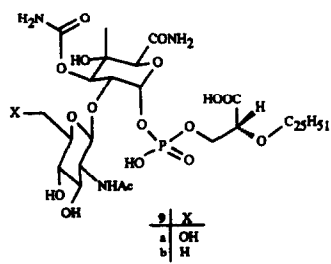
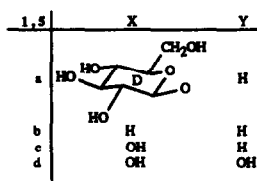
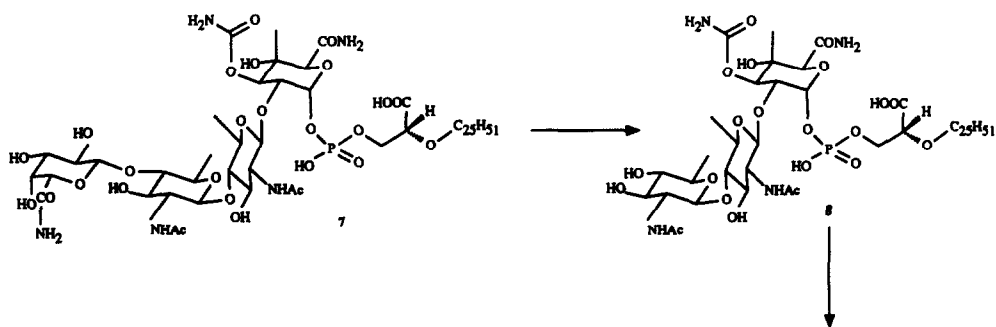
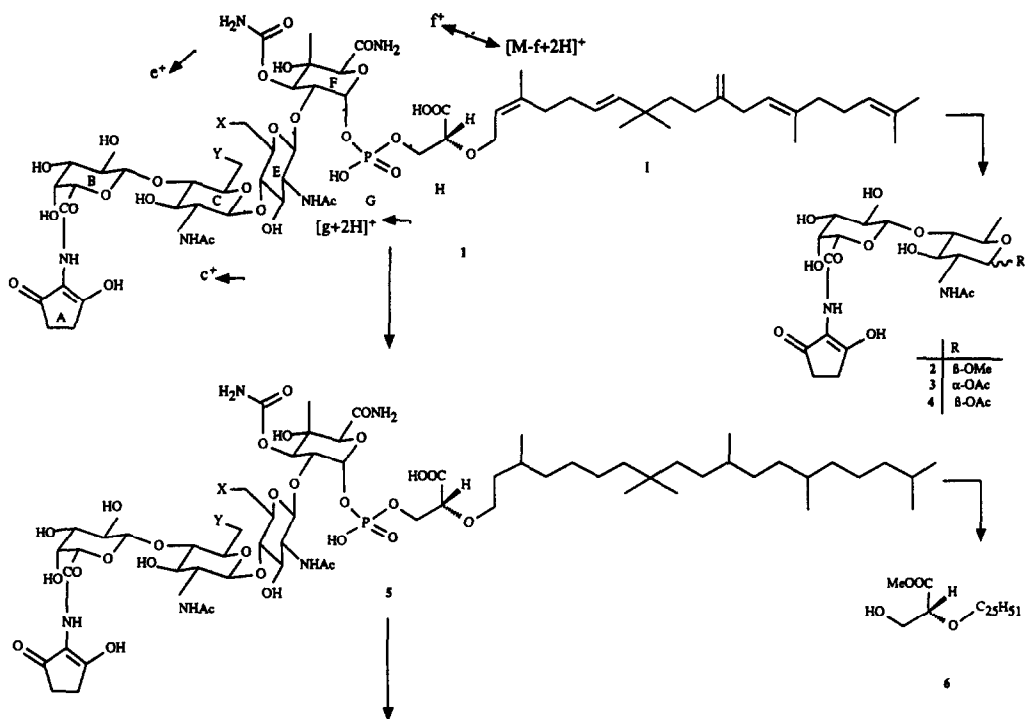
(Received in Germany 8 January 1993)

Abstract - Isolation, structural assignment, and antibiotic efficiency of the new moenomycin antibiotics C₃ (1b) and C₄ (1c) is described. The previously published structure of pholipomycin (1d) is modified.

Introduction

The family of moenomycin-type antibiotics includes the components of the flavomycin^R complex, the prasinomycins, diumycins (marcarbomycins), 11837 R.P., 8036 R.P. (quebemecin), 19402 R.P., ensachomycin, prenomycin, teichomycin, and pholipomycin.¹ They are very efficient inhibitors of the bacterial cell wall peptidoglycan biosynthesis (*in vivo* activity mainly against *gram*-positive bacteria). Work by van Heijenoort and coworkers has demonstrated that moenomycin A interacts with the enzyme(s) that catalyses the formation of the linear peptidoglycan strands from a membrane disaccharide phospholipid intermediate.² Structurally, the moenomycin-type antibiotics are rather complex. Until now, only the structure of moenomycin A (1a) seems to be firmly established.³ Practically nothing is known on the biosynthesis of 1a and related antibiotics. We reasoned that from the structures of further members of the family at least some hints might be gained on how the rather complicated array of building blocks is assembled in the course of the biosynthesis. Even in the days of highly developed tools, the structure elucidation of these compounds is by no means simple. X-ray analysis cannot be used, since (at the best of our knowledge) none of the moenomycin antibiotics has ever been obtained crystalline. More importantly, ¹H NMR spectroscopy is usually of little help. In most instances very broad peaks are observed, probably the result of the well-known association of these compounds in aqueous solution¹ and of other time-dependent phenomena.

In the present paper we describe the structure elucidation of the new flavomycin^R component moenomycin C₃ and propose a structure for moenomycin C₄. Included is some work on pholipomycin. The structure of this antibiotic was previously elucidated, relying on the moenomycin A work.⁴ In view of the recent



modification of the moenomycin A structure³ (unit F, see formula 1a, is derived from the amide of moenuronic acid rather than from the free acid, as originally assumed) it seemed probable that the structure of pholipomycin also needed some modification

Isolation of Moenomycins C₃ and C₄

TLC plates of the trade product flavomycin^R (solvent system 1-propanol - 2N NH₃ 70/30) show two spots for moenomycin A (1a) and the less polar moenomycin C (ratio 3:1).⁵ Reversed-phase HPLC has now demonstrated moenomycin C to be a mixture of several components. Careful chromatographic separation of this mixture has led to the isolation of two new antibiotics which were named moenomycin C₃ and C₄ (according to their retention times⁶).

Spectral Properties of Moenomycin C₃

A careful analysis of the positive ion FAB mass spectra of moenomycin A (1a) and degradation products derived thereof has revealed that all structurally relevant fragments can be assigned as summarized in formula 1. Cleavage of the glycosidic bonds of the pyranose units C, E, and F gives rise to the formation of cations c⁺, e⁺, f⁺, stabilized by the respective pyranose oxygens. Cleavage of either phosphoric acid diester bond yields the protonized phosphoric monoesters [M-f+2H]⁺ and [g+2H]⁺, respectively.³ Application of these rules to moenomycin C₃ clearly indicated its structure to differ from that of moenomycin A by the absence of both the glucose unit D and the OH group at C-6 of the amino sugar moiety E, see formula 1b. The ¹³C NMR spectrum of moenomycin C₃ was very informative showing the presence of the moenocinol unit I, the chromophore part A, the carbamoyl group, four sugar units (well separated anomeric carbon signals), two of them being 2-N-acetylamino-2-deoxy sugars (C-2 signals at $\beta = 56.2$ and 57.2), and the ³¹P, ¹³C coupling in the vicinity of the phosphate group. The assignments are collected in the Experimental

Chemical Degradation of Moenomycin C₃ (1b)

Treatment of moenomycin C₃ with trifluoroacetic acid (5h at 60°C), a method known to cleave the β -glycosidic bonds of 2 α -acetylamino-2-deoxy sugars rather specifically,⁷ followed by (i) trapping the intermediate oxazolium ion with methanol, (ii) acetylation, and (iii) chromatographic separations furnished 2 along with a 3:2 mixture (¹H NMR) of 3 and 4, from which pure 3 was obtained by crystallization. The structure of these compounds was rigorously established by comparison with the products that were obtained when moenomycin A (1a) was submitted to the same conditions.⁸

Hydrogenation of 1b in 33:1 methanol-acetic acid (Pt catalyst)⁹ provided decahydromoenomycin C₃ (5b) in 95% yield. Cleavage of 5b under rather vigorous conditions with trifluoroacetic acid (16.5 h at 60°C), followed by ester formation with 1.5% HCl in methanol, and separations provided the known compound 6¹⁰ in 47% overall yield. This part of the work proves, that moenomycin C₃ contains units A-B-C and H-I.

5b was then submitted to the stepwise degradation developed for 5a.⁹ Thus, oxidation of 5b with K₃[Fe(CN)₆] in 0.37 mol/l K₂CO₃ solution removed unit A selectively to give 7 in 53% yield. From 7 the galacturonic acid moiety B was degraded by (i) diol cleavage with NaIO₄ in 50% acetic acid containing sodium acetate (5h at 40°C), (ii) reversed-phase chromatography, (iii) treatment with N,N-dimethylhydrazine in 2-propanol - 1 mol/l H₂SO₄ (3h at 85°C), and (iv) chromatographic separations to furnish 8 in 54% yield. A second degradation cycle in which ammonia was used instead of N,N-dimethylhydrazine converted 8 into 9b (33%).

Spectroscopic Properties of the Moenomycin C₃ Degradation Products

Positive ion FAB spectra of compounds **5b**, **7**, **8**, and **9b** have been obtained fully in accord with the proposed structures (see Experimental). The ¹³C chemical shifts for compounds **7**, **8** and **9b** (see Table 1) were assigned making use of the DEPT sequence and by comparison with known moenomycin A degradation products. The presence of two 6-deoxy sugars in **7** carrying ether substituent at C-4 is indicated by the two low-field signals at $\delta = 86.1$ and $\delta = 84.3$.¹¹ In **8** one and in **9b** both these signals are absent. This result seems only to be consistent with a 1 \rightarrow 4 linkage between units C and E. For the first time under carefully selected experimental conditions (solvent CDCl₃-CD₃OD-D₂O 18:11:2:7) informative ¹H NMR spectra of moenomycin-type compounds have been obtained. The results are collected in Table 2. Assignments are based on H/H COSY. From the coupling between the ring protons of unit E (see Table 2) gluco configuration, ⁴C₁(D) conformation and a β -glycosidic linkage to unit F may be deduced.¹² Similarly, $J = 8.3$ Hz for the coupling between 1-H and 2-H of the quinovosamine unit C in **8** is indicative of a β -glycosidic bond. From $J_{2,3}(F) \approx 10.5$ Hz it may be concluded that the moenuronic acid moiety F also adopts the ⁴C₁(D) conformation.

Table 1 ¹³C NMR spectral data of compounds **7**, **8**, and **9b** (δ values)

7a	8b	9bb	assignment	unit
173.4	-	173.1	C-1	H
172.1	172.2	172.3	C-6	F
171.1	172.1	-	NHCOCH ₃	C
170.5	-	-	C-6	B
170.0	172.0	172.3	NHCOCH ₃	E
156.9	157.2	?	OCONH ₂	F
103.8	-	-	C-1	B
102.1	101.7	102.0	C-1	E
101.5	101.7	-	C-1	C
93.8	94.6	94.6	C-1	F
86.1	85.8	75.1	C-4	E
84.3	74.9	-	C-4	C
80.8	77.6	77.8	C-2	H
76.0	76.7	75.7	C-2	F
74.6	73.4	-	C-3	C
74.3	-	-	C-3	B
74.3	74.1	74.3	C-3	F
72.9	72.4	72.5	C-4	F
72.5	71.7	71.5	C-5	E
72.1	72.0	72.0	C-5	F
71.7	71.6	73.3	C-3	E
70.8	-	-	C-5	B
70.2	-	-	C-2	B
69.8	-	-	C-4	B
68.9	69.7	-	C-5	C
67.5	69.3	69.3	C-1	I
66.3	66.0	66.0	C-3	H
55.8	55.3	-	C-2	C
55.2	54.8	55.5	C-2	E
?	?	36.0	C-2	I
24.5	22.0	22.0	NHCOCH ₃	E
24.4	21.7	21.9	NHCOCH ₃	C
?	12.2	16.7	C-6	E

^a in DMSO-d₆ ^b in CDCl₃-CD₃OD-D₂O 18:11:2:7 + trifluoroacetic acid (6 μ L/mL)

Table 2. ^1H NMR spectral data of **8**, **9b**, and moenomycin A degradation product **9a** (δ and J values)^a

8	9b	9a	assignment	unit
4 17			1-H	C
3 43			2-H	C
3 12			3-H	C
2 83			4-H	C
3 10			5-H	C
1 10			CH ₃ -6	C
1 72			NHCOCH ₃	
$J_{1,2}=8$ 3Hz $J_{2,3}=10$ 5Hz $J_{3,4}=8$ 9Hz $J_{4,5}=8$ 9Hz				
4 17	4 17	4 20	1-H	E
3 39	3 36	3 25	2-H	E
3 28	3 16	3 12	3-H	E
2 86	2 82	2 98	4-H	E
3 14	3 06	2 95	5-H	E
-	-	3 37, 3 50	6-H, 6-H'	-
?	1 05	-	CH ₃ -6	E
1 72		1 67	NHCOCH ₃	
$J_{1,2}=8$ 3Hz $J_{3,4}/J_{4,5}=8$ 0Hz/ 10 0Hz				
$J_{1,2}=8$ 3Hz $J_{2,3}=9$ 0Hz $J_{3,4}=9$ 0Hz $J_{4,5}=9.0$ Hz				
$J_{1,2}=8$ 6Hz $J_{2,3}=10$ 0Hz $J_{3,4}=10$ 0Hz $J_{4,5}=10$ 0Hz $J_{5,6'}=5$ 8Hz $J_{5,6}=2$ 0Hz $J_{6,6'}=13$ 0Hz				
5 49	5 47	5 63	1-H	F
3 31	3 33	3 33	2-H	F
4 75	4 75	4 69	3-H	F
4 08	4 10	4 00	5-H	F
?	0 96	?	CH ₃ -6	F
$J_{2,3}=10$ 5Hz	$J_{2,3}=10$ 8Hz	$J_{1,2}=3$ 5Hz $J_{1,P}=8$ 0Hz $J_{2,3}=10$ 8Hz		
3 83	3 83	3 75	2-H	H
3 89-3 97	3 91,3 94	3 82,3 86	CH ₂ -3	H
$J_{2,3}=3$ 5Hz $J_{2,3'}=8$ 0Hz				
3 16, 3 20	3 25, 3 39	3 17, 3 31	CH ₂ -1	I
1 03, 1 07	1 27, 1 39	1 10, 1 32	CH ₂ -2	I

^ain CDCl₃-CD₃OD-D₂O 18 11 2 7

Moenomycin C₄

From the FAB MS (for details, see Experimental) it was obvious that moenomycin C₄ contained one more oxygen than moenomycin C₃ (1b). Fragmentation e (see formula 1) led to an ion that was 16 mass units larger than in the case of 1b whereas fragmentation c was the same as found previously in moenomycin A (1a). Thus the extra oxygen (with respect to 1b) is located in unit E. We propose, therefore, structure 1c for moenomycin C₄. This structure is well in accord with the ¹³C NMR spectrum (see Experimental), but it has not rigorously been proven.

Pholipomycin

As was already mentioned above, in view of the recent structural revision of moenomycin A it seemed very probable that unit F of pholipomycin also contains the amide of moenuronic acid rather than the free uronic acid. And indeed, a careful mass determination of the M+K ion of both pholipomycin and its decahydro derivative has unambiguously shown that the molecular mass of pholipomycin is one mass unit less than the structure with moenuronic acid in its free acid form requires. Thus, as in the case of all the other moenomycins, pholipomycin has a uronamide unit F and should be represented by formula 1d.

Biological activity of moenomycins C₃ and C₄

The minimum inhibitory concentrations (MIC) of moenomycins C₃ (1b) and C₄ against various microorganisms have been determined by a serial two-fold agar dilution method (Müller Hinton Agar). In addition, inhibition of the UDP-N-acetylmuramyl pentapeptide-dependent incorporation of [¹⁴C]UDP-N-acetylglucosamine into cross-linked high-molecular weight peptidoglycan was studied with a slightly modified¹³ version of the assay described by Izaki, Matsuhashi, and Strominger¹⁴.

Table 3 Minimal inhibitory concentrations (in g/L) of moenomycins C₃ (1b), C₄ (1c), and A (1a, for comparison) against various test organisms

test organism	1b	1c	1a
Staph aureus SG 511	0.10	0.05	0.05
Strept pyogenes 77	< 0.01	< 0.01	< 0.01
Strept pyogenes 308	25	6.25	25
Pseud aerug 1771	50	25	100
Pseud aerug 1771M	1.56	1.56	3.13
E coli DC 2	25	25	50
Salm typhimurium	100	100	> 100
Klebs aerog 1082E	100	100	> 100
Ent cloacae 1321E	1.56	0.78	1.56

Table 4 Effect of moenomycins C₃ (1b), C₄ (1c), and A (1a, for comparison) on the *in-vitro* UDP-N-acetylmuramyl pentapeptide-dependent incorporation of [¹⁴C]UDP-N-acetylglucosamine into cross-linked high-molecular weight peptidoglycan

concentration (mg/L)	% inhibition		
	1b	1c	1a
1	85	85	88
0.1	63	60	63
0.01	0	0	0

Finally, the inhibitory effect of 1b and 1c directly on the transglycosylation reaction was determined by the *in vitro* assay developed earlier in one of our laboratories¹⁵ which uses a particulate fraction from *E. coli* and as substrate the lipid intermediate which is the immediate precursor of uncross-linked peptidoglycan. The results are summarized in Tables 3 - 5.

Table 5 Effect of moenomycins A (1a, for comparison), C₃ (1b) and C₄ (1c) on the *in-vitro* formation of peptidoglycan by transglycosylation^a

final concentration (μg/mL)	% inhibition		
	1b	1c	1a
10	100	100	100
1	100	100	100
0.1	91	86	78

^aAssays were carried out as described previously¹⁵

Discussion

Like moenomycin A (1a), the new moenomycins 1b and 1c are *in-vitro* highly active only against gram-positive bacteria (see Table 3). As far as their inhibitory effect on the transglycosylation reaction is concerned, 1a, 1b, and 1c possess practically the same inhibitory efficiency (Table 5). This is well in agreement with the previously established structure-activity relations which state, that only units E-F-G-H-I are essential for full transglycosylase inhibiting activity. From the results collected in Table 4 one may arrive at the same conclusion.

The structures of the moenomycin antibiotics described above seem to shed some light on biosynthetic events. Obviously, first the whole complicated structure is assembled and is subsequently modified. One is tempted to speculate that pholipomycin (1d) is the precursor of the other members of the group which could be formed from 1d by glucosidation and deoxygenations, respectively.

EXPERIMENTAL

General

O₂- or moisture-sensitive reactions were performed in oven-dried glassware under a positive pressure of argon. Liquids and solutions were transferred by syringe. Small-scale reactions were performed in Wheaton serum bottles sealed with aluminium caps with open top and Teflon-faced septum (Aldrich). Solvent evaporations were performed in vacuo at 40°C using a rotatory evaporator, Lyophilizations Leybold-Heraeus GT2. Solvents were purified by standard techniques. The instrumentation used was ¹H NMR: WP 80 (Bruker), AM 400 (Bruker), ¹³C NMR: AM 400 (Bruker) at 100.6 MHz, EI MS MAT CH5 (Varian), FAB MS (i) MAT 731 (Varian) with a modified Saddle Field Source, (ii) VG AUTOSPEC, (iii), VG Analytical ZAB2-SEQ (BEQQ configuration), LC (preparative gravitational liquid chromatography) silica gel (ICN Biomedicals Silica 63-100), MPLC (medium-pressure liquid chromatography): 40.0 cm x 4.5 cm glass tubes, 50 μm silica gel (Amicon), Duramat pump (CfG), Thomachrom UV detector (Reichelt), analytical TLC Merck precoated silica gel 60 F₂₅₄ plates (0.2 mm), spots were identified under a UV lamp (Camag 29 200) and by spraying with a 2.22 mol/l H₂SO₄ solution which contained Ce(SO₄)₂·xH₂O (10 g/l) and H₃[PO₄(Mo₃O₉)₄]_xH₂O (25 g/l)¹⁶ and heating at 140°C. For crude reversed-phase separations polystyrene resin HP-20 (Mitsubishi) was used. Carbon and proton numbering in the subunits (see NMR data) follows the moenomycin nomenclature (see formula 1). Two molecular masses are always communicated, the first refers to ¹²C, ¹H, ¹⁶O, ¹⁴N, ³¹P (mono-isotopic masses), the second was calculated using the International Atomic Masses.

Isolation of moenomycins C₃ and C₄

Silica gel chromatography (300g SiO₂, Grace 60-200 μm) of moenomycin complex (Flavomycin^R, 12.0 g) performed as described previously¹⁷ provided moenomycin C complex (1.05 g) and pure moenomycin A (1.87 g). A moenomycin C mixture (1.35 g) was separated by preparative HPLC (Waters prep LC 500, Merck LiChroprep RP-18, 25-40 μm, mobile phase methanol-acetonitrile-water 52:8:40, flow rate 25 mL/min). First the column was washed with 1 l of the solvent mixture then 15 mL fractions were taken. Fractions 52-68 contained moenomycin C₃ (130 mg) and fractions 69-80 (173 mg) moenomycins C₃ (55%) and C₄ (38%). The latter fraction was further purified under the same conditions and furnished pure moenomycin C₄. Analytical HPLC Spherisorb ODS 5 μm, solvent system methanol-acetonitrile-0.02% phosphate buffer (pH 7.8) 4:1:5, UV detection at 258 nm.

Moenomycin C₃ (1b)

UV λ_{max} (ε) in methanol 259 (21000), in methanol + HCl 243 (12000) - ¹³C NMR ((100.6 MHz, D₂O-CD₃OD 1:1, DEPT) δ = 16.10 (CH₃), 16.29 (CH₃), 17.80 (CH₃), 17.84 (CH₃), 17.95 (CH₃), 23.15 (CH₃), 23.20 (CH₃), 23.33 (CH₃), 23.38 (CH₃), 23.97 (CH₃), 25.96 (CH₃), 27.59 (CH₂), 27.82 (CH₃, C-23^I, C-24), 31.47 (CH₂), 32.21 (CH₂), 32.60 (CH₂), 33.40 (CH₂), 35.89 (C-12^I), 36.39 (C-8^I), 40.80 (C-15^I), 42.77 (C-9^I), 56.24 (C-2^E), 57.19 (C-2^C), 67.04 (C-1^I), 68.63 (C-3^H), 70.66 (CH), 71.44 (CH), 72.28 (CH), 72.65 (CH), 73.48 (CH), 73.86 (CH), 74.22 (CH), 75.99 (CH), 76.52 (CH), 78.37 (C-2^F), 80.51 (C-2^H), 84.60 (C-4^C), 87.72 (C-4^E), 95.91 (C-1^F), 103.42 (C-1^C), 103.93 (C-1^E), 104.70 (C-1^B), 109.27 (C-22^I), 111.29 (C-2^A), 123.09 (C-13^I), 123.44 (C-2^I), 125.26 (C-17^I), 126.83 (C-6^I), 132.24 (C-18^I), 137.30 (C-14^I), 141.23 (C-3^I), 141.45 (C-7^I), 151.02 (C-11^I), 159.20 (OCONH₂), 159.23, 170.23 (C-6^B), 173.82 (C-6^F), 173.92, 174.46 (NHCOCH₃), 174.56 (NHCOCH₃), 177.05 (C-1^H), 199.54 (C-1^A, C-3^A) - C₆₃H₉₈N₅O₂₈P (1403.6, 1404.5), FAB MS (matrix lactic acid) 1480.4 ([M+2K-H]⁺), 1442.4 ([M+K]⁺), 916.2 ([f-H+K]⁺), 684.1 ([e-H+K]⁺)

Decahydromoenomycin C₃ (5b)

1b (51.2 mg, 36 μmol) and PtO₂ (12.8 mg) in 30 l methanol-acetic acid (5.2 mL) were stirred under hydrogen for 3 d at 20°C. Progress of the reaction was monitored by HPLC (5 μm RP-18, methanol-water-acetonitril (6:3:1), and by TLC (chloroform-methanol-water 18:11:2:7). After filtration, solvent evaporation and lyophilization the reaction product was purified by reversed-phase LC (20 g HP 20, gradient water - methanol) to give pure 5a (49.1 mg, 95%) - δ = 56.38 / 57.41 (C-2^E and C-2^C), 68.52 (C-1^I), 70.14 (C-3^H), 70.70, 71.59, 72.41, 72.75, 73.58, 73.78, 73.98, 74.21, 74.47, 75.92, 76.77, 78.60, 84.74 (C-4^C), 87.80 (C-4^E), 96.12 (C-1^F), 103.55 / 104.11 / 104.80 (C-1^C, C-1^E, C-1^B), 111.62 (C-2^A), 159.16 (OCONH₂), 170.21 (C-6^B), 173.60, 174.20, 174.75 (2xNHCOCH₃, C-6^F, C-1^H), 198.92 C-1^A, C-3^A) - C₆₃H₁₀₈N₅O₂₈P (1413.7, 1414.5), FAB MS (matrix methanol - nitrobenzylalcohol) 1458.7 ([M+2Na-H]⁺), 1436.6 ([M+Na]⁺), 1020.3 ([g+2Na]⁺), 900.3 (f+Na-H)⁺, 668.3 ([e+Na-H]⁺), 559.4 ([M-f+Na+H]⁺), 459 ([c]⁺)

Methyl (R)-2-(3,8,8,11,14,18-hexamethylnonadecyloxy-3-hydroxypropionate) (6)

In a 1 mL sealed vessel a mixture of 5b (49.1 mg, 34.7 μmol) and trifluoroacetic acid (TFA, 0.25 mL) was stirred at 60°C for 16.5 h. After TFA removal (stream of argon) the dark brown residue was treated with 1.5 per cent HCl in methanol at 70°C for 4h. After dilution with CH₂Cl₂, and solvent evaporation the residue was taken up in CH₂Cl₂ (6 mL). Washing with 5 per cent aq NaHCO₃ solution (2 x 1 mL), H₂O (2 x 1 mL), drying, and solvent evaporation, followed by LC (2.5 g SiO₂, CHCl₃-MeOH 100:1) of the residue (13.4 mg) yielded pure 6 (7.6 mg, 46%), identical with a sample obtained from 1a - ¹H NMR (80 MHz, CDCl₃) δ = 0.80-1.60 (signals of unit I), 3.80 (s, OCH₃), 3.82-4.10 (2-H^H, CH₂-3^H, CH₂-1^H) - C₂₉H₅₈O₄ (470.4, 470.8), EI-MS 470 (M⁺), 452 ([M-H₂O]⁺), 440 ([C₂₈H₅₆O₃]⁺), 411 ([M-COOCH₃]⁺), 349 ([C₂₅H₄₉]⁺), 273 ([M-C₁₄H₂₉]⁺), 239, 238, 169, 121 ([C₄H₉O₄]⁺)

Degradation of 1b with trifluoroacetic acid¹⁹

a) A mixture of 1b (300.1 mg, 0.21 mmol) and trifluoroacetic acid (1.5 mL) was heated to 60°C for 5h in a sealed vessel. Most of the trifluoroacetic acid was then removed by blowing argon over the mixture. The residue was dissolved in dry methanol and the solution was allowed to stand at 40°C for 17 h. After solvent evaporation and LC (25 g SiO₂, chloroform-methanol-water 18:11:2:7) the fractions containing compounds with R_f values in the range of 0.1 (287.6 mg, TLC chloroform-methanol-water 18:11:2:7) were collected, evaporated to dryness, and the residue was dissolved in 1 l pyridine-acetic anhydride (6 mL) and left at 20°C for 24 h. Solvent evaporation, lyophilization, and MPLC (column B, chloroform-methanol-water 30:5:0:5) gave a fraction (21.2 mg) which according to the 400 MHz ¹H NMR spectrum was a mixture of 3 and 4, and a fraction consisting of the more polar compound 2 (27.4 mg). Both fractions were rechromatographed under the same conditions (using column A) 3 and 4 which could not be separated chromatographically but yielded pure 3 (10.4 mg) on crystallization from hexanes-ethyl acetate.

b) Moenomycin A (1a) was degraded under the same conditions, to give 2 and 3 which were identical with the samples obtained from 1b

Methyl 4-O-[2,3,4-tri-O-acetyl-β-D-galactopyranosyluron-N-(1-hydroxy-3-oxo-1-cyclopenten-2-yl)amide]-(1-4)-2-acetamido-3-O-acetyl-2,6-dideoxy-β-D-glucopyranoside (2)

M p. of the sample obtained from moenomycin C₃ (1b) 241-243°C (from hexanes-ethanol), m p. of the sample obtained from 1a 241-243°C (from hexanes-ethanol) - For ¹H NMR and mass spectra, see ref 8a, 8b

O-2,3,4-Tri-O-acetyl-β-D-galactopyranosyluron-N-(1-hydroxy-3-oxo-1-cyclopenten-2-yl)amide -(1-4)-2-acetamido-1,3-di-O-acetyl-2,6-dideoxy-α-D-glucopyranose (3)

M p. of the sample obtained from moenomycin C₃ 265-267°C (from hexanes-ethanol), m p. of the sample obtained from 1a 265-266°C (from hexanes-ethanol) - For ¹H NMR and mass spectra, see ref 8a, 8b

2-O-[2-Acetylamino-4-O-[2-acetylamino-4-O-((5S)-5-carbamoyl-β-L-arabinopyranosyl)-2,6-dideoxy-β-D-glucopyranosyl]-2,6-dideoxy-β-D-glucopyranosyl]-3-O-carbamoyl-1-O-[[S]-2-carboxy-2-(3,8,8,11,14,18-hexamethyl-nonadecyloxy)-ethoxyl-hydroxyphosphoryl]-4-C-methyl-α-D-glucopyranuronamide (7)

5b (362.3 mg, 0.26 mmol) was degraded with K₃[Fe(CN)₆] (841.2 mg, 2.49 mmol) essentially as described in ref 9 for 5a. MPLC (2 x column B, chloroform-methanol-water 18:11:2:7) gave pure 7 (179.0 mg, 53%) - ¹³C NMR (62.9 MHz, DMSO-d₆, DEPT) δ = 24.4 (NHCOCH₃), 24.5 (NHCOCH₃), 55.2 (C-2E), 55.8 (C-2C), 66.3 (C-3H), 67.5 (C-1), 68.9 (C-5C), 69.8 (C-4B), 70.2 (C-2B), 70.8 (C-5B), 71.7 (C-3E), 72.1 (C-5F), 72.5 (C-5E), 72.9 (C-4F), 74.3 (C-3B, C-3F), 74.6 (C-3C), 76.0 (C-2F), 80.8 (C-2H), 84.3 (C-4C), 86.1 (C-4E), 93.8 (C-1F), 101.5 (C-1C), 102.1 (C-1E), 103.8 (C-1B), 156.9 (OCONH₂), 170.0 (NHCOCH₃), 170.5 (C-6B), 171.1 (NHCOCH₃), 172.1 (C-6F), 173.4 (C-1H) - C₅₈H₁₀₄N₅O₂₆P (1317.7, 1318.5), FAB MS (glycerol) 1385 ([M+3Na-2H]⁺), 1379 ([M+Na+K-H]⁺), 1363 ([M+2Na-H]⁺), 1341 ([M+Na]⁺), 1319 ([M+H]⁺), 804 ([f+Na-H]⁺), 597 ([M-f+Na+K]⁺), 581 ([M-f+2Na]⁺), 572 ([e+Na-H]⁺), 559 ([M-f+Na+H]⁺)

2-O-[2-Acetylamino-4-O-[2-acetylamino-2,6-dideoxy-β-D-glucopyranosyl]-2,6-dideoxy-β-D-glucopyranosyl]-3-O-carbamoyl-1-O-[[S]-2-carboxy-2-(3,8,8,11,14,18-hexamethyl-nonadecyloxy)-ethoxyl-hydroxyphosphoryl]-4-C-methyl-α-D-glucopyranuronamide (8)

7 (189.8 mg, 0.14 mmol) was degraded with (i) NaIO₄ and then with dimethylhydrazine exactly as described in ref 9. Two chromatographic separations: a) LC (8 g SiO₂, chloroform-methanol-water 10:6:1), b) MPLC (column B, the same solvent system) gave pure 8 (89.7 mg, 54%) - C₅₂H₉₅N₄O₂₁P (1142.6, 1143.3), FAB MS (TEA) 1204 ([M+Na+K-H]⁺), 1188 ([M+2Na-H]⁺), 1182 ([M+K]⁺), 1166 ([M+Na]⁺), 1144 ([M+H]⁺), 629 ([f+Na-H]⁺), 597 ([M-f+Na+K]⁺), 581 [M-f+2Na]⁺, 559 ([M-f+Na+H]⁺), 397 ([e+Na-H]⁺), 375 ([e]⁺)

2-O-(2-Acetylamino-2,6-dideoxy-β-D-glucopyranosyl)-3-O-carbamoyl-1-O-[[S]-2-carboxy-2-(3,8,8,11,14,18-hexamethyl-nonadecyloxy)-ethoxyl-hydroxyphosphoryl]-4-C-methyl-α-D-glucopyranuronamide (9b)

8 (89.0 mg, 77.9 μmol) was degraded with (i) NaIO₄, (ii) 25 per cent aqueous NH₃ exactly as described in ref 9. MPLC (column A, chloroform-methanol-water 18:11:2:7) provided pure 9a (24.9 mg, 33%) - C₄₄H₈₂N₃O₁₇P (955.5, 956.1), FAB MS (matrix nitrobenzylalcohol) 1022.5 ([M+3Na-2H]⁺), 1000.6 ([M+2Na-H]⁺), 581 (M-f+2Na)⁺, 559.3 ([M-f+Na+H]⁺), 442.2 ([f+Na-H]⁺).

Moenomycin C₄ (1c)

UV λ_{max} (ε) in methanol + HCl 243 nm (12000) - ¹³C NMR (100.6 MHz, CD₃OD at 50°C, DEPT) δ = 16.16 (CH₃), 16.39 (CH₃), 17.76 (CH₃), 18.15 (CH₃), 23.20 (CH₃), 23.34 (CH₃), 23.93 (CH₃), 25.84 (CH₃), 27.74 (CH₂), 27.89 (CH₃, C-23), C-24), 31.68 (CH₂), 32.37 (CH₂), 32.59 (CH₂), 33.53 (CH₂), 35.97 (C-12), 36.42 (C-8), 40.79 (C-15), 42.94 (C-9), 56.20 (C-2E), 57.49 (C-2C), 61.06 (C-6C), 67.07 (C-1), 68.28 (C-3H), 70.78 (CH), 71.86 (CH), 72.40 (CH), 72.78 (CH), 73.82 (CH), 74.08 (CH), 74.52 (CH), 76.14 (CH), 76.78 (CH), 78.11 (CH), 79.56 (C-2F), 81.55 (C-2H), 84.93 (C-4C), 95.91 (C-1F), 103.14 (C-1C), 104.38 (C-1E), 104.82 (C-1B), 109.24 (C-22), 111.55 (C-2A), 123.15 (C-13), 123.49 (C-2), 125.38 (C-17), 126.86 (C-6), 132.17 (C-18), 137.33 (C-14), 141.62 (C-3), C-7), 151.15 (C-11), 153.50 (OCONH₂), 169.85 (C-6B), 173.80 (C-6F), 174.34 (2xNHCOCH₃), 177.62 (C-1H), 192.44 (C-1A, C-3A) - C₆₃H₉₈N₅O₂₉P (1419.6, 1420.5), FAB MS (matrix lactic acid) 1496.3 ([M+2K-H]⁺), 1458.4 ([M+K]⁺), 932.1 ([f-H+K]⁺), 700.1 ([e-H+K]⁺), 603.1 ([M-f+2K]⁺), 565.2 ([M-f+H+K]⁺), 497.0 ([c-H+K]⁺), 459.1 ([c]⁺), 417 ([c-42]⁺).

Polipomycin (1d)

1d was purified by preparative HPLC (10 μ m RP 18, methanol-acetonitrile-0.02% phosphate buffer (pH 7.8) 10:2:5, followed by LC (HP 20 (17g), elution with water and methanol) The methanol fraction contained 1d.- ^{13}C NMR ((100.6 MHz, CD_3OD) δ = 16.12, 16.37, 17.79, 23.11, 23.30, 23.97, 25.95, 27.72, 27.86, 30.44, 30.68, 30.77, 31.70, 32.34, 32.70, 33.54, 35.94, 36.46, 40.89, 42.89, 56.24 / 57.22 (C-2E and C-2C), 61.21 / 61.51 (C-6C and C-6E), 67.11, 68.92, 70.80, 72.17, 73.59, 74.07, 74.20, 74.33, 74.40, 75.97, 76.03, 76.91, 76.96, 79.13, 79.56, 79.64, 80.92, 81.23, 96.03 (C-1F), 103.32 / 104.15 / 104.49 (C-1C, C-1E, C-1B), 109.24 (C-22), 111.18 (C-2A), 123.50, 125.36, 126.94, 132.20, 137.33, 140.76, 141.48, 151.08, 159.43 (OCONH₂), 170.12, 173.80, 174.42, 174.51, 177.70, 200.00 (C-1A, C-3A).- $\text{C}_{63}\text{H}_{98}\text{N}_5\text{O}_{30}\text{P}$ (1435.6, 1436.5), FAB MS (narrow voltage scan, resolution 7300, internal reference: PEG-1540, matrix: nitrobenzylalcohol) m/z (mean of 3 measurements, $S \pm 0.002$) = 1474.57 (IM + K)⁺, calc 1474.57)-

Decahydropholipomycin (5d)

To 1d (13.0 mg) and PtO_2 (4 mg) were added methanol (1.5 mL) and acetic acid (49 μ L), and the mixture was stirred under H_2 for 135 min. Filtration, solvent evaporation, and preparative HPLC (RP 18, methanol-water-acetonitril 6:2:1) furnished 5d (9.4 mg) - $\text{C}_{63}\text{H}_{108}\text{N}_5\text{O}_{30}\text{P}$ (1445.7, 1446.5), FAB MS (conditions as described for 1d): 1484.65 (IM + K)⁺, calc 1484.65)

Acknowledgments - The group at Bochum kindly acknowledges financial support by the Fonds der Chemischen Industrie and the Hoechst AG

REFERENCES AND NOTES

- ¹Review: Huber, G. in Hahn, F.E. (ed.) *Antibiotics*, vol V/1, p. 135-153, Springer, Berlin 1979
- ²Review: van Heijenoort, J.; van Heijenoort, Y.; Welzel, P. in Actor, P.; Daneo-Moore, L.; Higgins, M.L.; Salton, M.R.J.; Shockman, G.D. (eds.) *Antibiotic Inhibition of Bacterial Cell Wall Surface and Function*, American Society for Microbiology, Washington, 1988, p. 549-557.
- ³Fehlhaber, H.-W.; Girg, M.; Seibert, G.; Hobert, K.; Welzel, P.; van Heijenoort, Y.; van Heijenoort, J. *Tetrahedron*, **1990**, *46*, 1557-1568.
- ⁴Takahashi, S.; Serita, K.; Arai, M.; Seto, H.; Furihata, K.; Otake, N. *Tetrahedron Lett.* **1983**, *24*, 499-502.
- ⁵The present *Streptomyces* strains used to manufacture flavomycin^R are the result of extensive strain selections. For the antibiotic formation of previously studied strains, see Schacht, U.; and Huber, G. *J. Antibiot.* **1969**, *22*, 597-602, and references therein
- ⁶The structure elucidation of moenomycin C₁ will be reported in a forthcoming publication.
- ⁷Welzel, P.; Knupp, G.; Wittler, F.-J.; Schubert, Th.; Duddeck, H.; Müller, D.; Höfle, G. *Tetrahedron* **1981**, *37*, 97-104.
- ⁸a) Welzel, P.; Wittler, F.-J.; Hermsdorf, L.; Tschesche, R.; Buhlke, H.; Michalke, P.; Simons, J.; Fehlhaber, H.-W.; Blumbach, J.; Huber, G. *Tetrahedron* **1981**, *37*, 105-112; b) Welzel, P.; Wittler, F.-J.; Hermsdorf, L.; Riemer, W. *Tetrahedron* **1981**, *37*, 113-118.
- ⁹Welzel, P.; Kunisch, F.; Krüggel, F.; Stein, H.; Scherkenbeck, J.; Hiltmann, A.; Duddeck, H.; Müller, D.; Maggio, J.E.; Fehlhaber, H.-W.; Seibert, G.; van Heijenoort, Y.; van Heijenoort, J. *Tetrahedron* **1987**, *43*, 585-598.
- ¹⁰Welzel, P.; Wittler, F.-J.; Müller, D. *Tetrahedron* **1976**, 1665-1668.
- ¹¹cf. ref ^{8b}
- ¹²Bock, K.; Thøgersen, H., in *Annual Reports on NMR Spectroscopy*, Vol. 13, p. 1-57, Academic Press, London, 1982.
- ¹³Schaller, K.; Hölte, J.-V.; Braun, V. *J. Bacteriol.* **1982**, *152*, 994-1000.
- ¹⁴Izaki, K.; Matsushashi, M.; Strominger, J.L. *J. Biol. Chem.* **1968**, *243*, 3180-3192
- ¹⁵van Heijenoort, Y.; Derrin, M.; van Heijenoort, J. *FEBS Lett.* **1979**, *89*, 141-144; van Heijenoort, Y.; van Heijenoort, J. *FEBS Lett.* **1980**, *110*, 241-244.
- ¹⁶Kritchewsky, D.; Kirk, M.R. *Arch. Biochem. Biophys.* **1952**, *35*, 346-351.
- ¹⁷Welzel, P.; Wietfeld, B.; Kunisch, F.; Schubert, Th.; Hobert, K.; Duddeck, H.; Müller, D.; Huber, G.; Maggio, J.E.; Williams, D.H. *Tetrahedron*, **1983**, *39*, 1583-1591
- ¹⁸Tschesche, R.; Blumbach, J.; Welzel, P. *Liebigs Ann. Chem.* **1973**, 407-418.
- ¹⁹Cf. ref ⁷